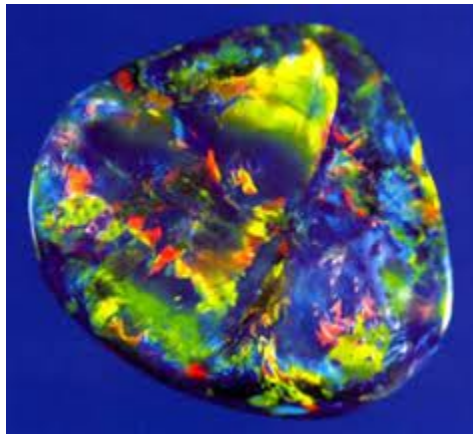




## Opal Facts: October's Birthstone

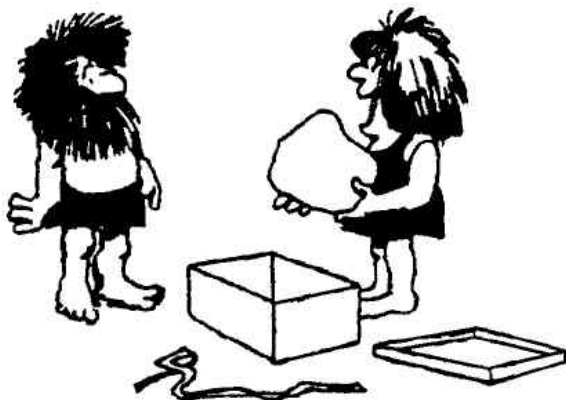
The name opal is derived from "Opalus" which meant "to see a change in color."



The 1<sup>st</sup> century Roman Pliny the Elder wrote "For in them you shall see the living fire of ruby, the glorious purple of the amethyst, the sea-green of the emerald, all glittering together in an incredible mixture of light". Romans popularized opals: an opal owned by Senator Nonius was so coveted by Mark Anthony that it is said he banished the senator for refusing to sell it to him. Estimated worth in today's currency-\$80,000. Since Roman times, opals were carried by their owners as a good luck charm or talisman of good fortune. Royalty prized opals: Queen Victoria wore opals and made them a Victorian fashion statement. Napoleon Bonaparte gave an opal to Josephine named 'the burning of Troy,' for its brilliant red flashes.

**From Australian Precious Opal: A Guide Book for Professionals by Andrew Cody, 1991 Mount Press Pty. Ltd, Melbourne, Australia**

Via The Rock Rattler Oct 2004



*"How Sweet... My Birthstone!"*

From Rock Chips 1/99

## Mexican Fire opal



Mexican Opal is considered a mineraloid and has no crystal structure. The term used for no structure is amorphous. Opal is scientifically written as  $\text{SiO}_2 \cdot \text{NH}_2\text{O}$ . It is called hydrated silicon dioxide. Stones that can commonly form Opal are limestone, basalt, rhyolite and sandstone. While Opal contains water, the water content does vary by stone. For instance, Opal has been found with water content approaching 20%; however, most Opal stones used for ornamental jewelry usually have water content of 3% to 10%. Fire opal, like all opal, has high water content. As a result, it should be protected from heat and prolonged exposure to strong light, which could dry it out. Dealers cure fire opal by drying it before cutting to make sure that any instability can be eliminated.

Mexican Opal is rated at 5.5 to 6.5 on Moh's Scale of hardness. The physical structure of Opal is unique. Tiny spheres of silicon dioxide form a pyramid shaped grid interspersed with water. Tiny natural faults in this grid cause the characteristic "play of color" - opalescence. The effect is similar to the rainbow colors displayed on a soap bubble, only much more dramatic. The incredible play of color is exhibited by the gemstone due to the presence of small spheres in the silica gel that causes interference and refraction appearance. Silicon and oxygen chains are packed in the spheres and these are uneven in size and incoherent in concentration. These are set in a compact structures thus, when ray of light falls it dissects the light on its way through the stone resulting in the awesome color play. As per the size of the spheres within the structure,

changeable colors of the gamut are diffracted. With the increase in the consistency of the spheres position, more intense and brilliant would be the color play.

Mexican opal that displays play of color but is rare because volcanic opal forms relatively quickly and the spheres of silica rarely have time to settle into the diffraction grids that create play of color.

Fire opal glows with the fire of the sun: hot honey yellows, oranges, and reds so bright they look as though they might glow in the dark. Fire opal sometimes does have opalescence or play of color like other opals but it does not need this to take a starring role in jewelry

Its juicy color is just the right accent to earth tones or black and also looks great paired with other bright tones

#### Cut



Fine Opal stones are cut and polished to round or oval cabochons or other softly domed shapes so as to display the best color. Cut is a very important characteristic of Opal, as the beauty of the crucial feature (i.e. color play) of the stone is dependent on its cut. The cut of the stone is the key to revealing its magnificence. First, the cutters remove the inclusions present in the stone with a diamond cutting wheel. Next a soft dome shape is given in order to bring out the fine color play. After cutting, the finishing is done using sandpaper, and finally the polishing is done with the help of a wet leather wheel. Jewelry designers can get many desirable cuts or the shapes from this striking stone. Oval, circular and round cabochons are preferred as in these shapes the color play is displayed at its best. Some other popular shapes are teardrop, square, rectangle and triangle etc.



Unlike most opal, fire opal is often faceted, so you can choose sparkle as well as color. Because it is light as well as bright, fire opal is especially good for earrings, where even small sizes have a big punch of color. Only the very finest qualities of Fire Opal are suitable for faceting.

#### Treatments

Opal is a soft stone and therefore is occasionally treated or impregnated with colorless oils, plastic, resin or wax which act as a bonding agent, reduce the visibility of the inclusions and improve the overall clarity of the stones. Black Opal is occasionally treated with chemicals, smoke, or dye. Opals are also sometimes backed with foil, black paint or lacquer. Thin opal is often made into doublets or triplets by covering the back and front.

Via The Pica Pick Sept 2011 Volume XXXVII Issue No. 1



## Bench Tips by Brad Smith

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### REMOVING A STONE FROM BEZEL SETTING

If you've forgotten to use dental floss and got your stone caught in a bezel, there's one thing you can try before starting to pry. Find some sticky wax or beeswax. Roll it into a pencil-sized cylinder and stick the end onto the top of the stone. Mold it on well and yank.

If all else fails, you either have to very carefully pry open the bezel with a sharp knife blade or drill a small hole in back of the stone and push it out with the point of a scribe.

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### MAKING YOUR OWN MOKUME

Ever think about making your own mokume? Here's a link to the detailed steps in the sequence as done by a professional. Look for mokume on <http://www.rchristopher.com/tech/>

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### FOREDOM STAND

A quick and easy way to suspend a Foredom over your jewelry bench is to use some steel pipe components from your local hardware store. It attaches with a couple screws and costs a little over \$10

I use ½ inch galvanized pipe and fittings. To build a stand that attaches to the top of your bench, all you'll need is a flange and a thirty inch length of the pipe. If you prefer a stand that attaches to the side of your bench, you'll need a little longer pipe, three foot, a flange, and a 90 degree "street elle".

Finally, make a hook that goes into the top of the pipe to hang the motor from. You can use heavy coat hanger wire or better yet, a 1/8 steel rod from the hardware store.

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Acknowledgement to be included with each  
publication:

More BenchTips by Brad Smith are at  
[groups.yahoo.com/group/BenchTips/](http://groups.yahoo.com/group/BenchTips/)  
or  
[facebook.com/BenchTips](http://facebook.com/BenchTips)

## SUMMER ROCKHOUND

By - Cindy Lind

The 'Skeeter and the Rockhound both  
Went out to hunt one day.  
The Rockhound wanted specimens,  
The 'Skeeter wanted prey!  
The Rockhound found a likely spot  
And settled with his gear,  
He'd dug a short time when he hears  
A buzzing in his ear.  
The Rockhound swatted at the noise  
And stood to look around.  
He missed the 'Sketter but he saw  
A great stone on the ground!  
The Rockhound bent to lift the stone  
His backside in the air.  
The 'Skeeter took advantage then  
And bit the Rockhound there.  
The Rockhound jumped and squashed the bug  
His bottom to defend.  
The Rockhound and the 'Skeeter had  
Both got it in THE END!

From Gulfport Gems 07 / 08 Via Beehive Buzzer 09 / 08  
Via THE CALGARY LAPIDARY JOURNAL OCT. 2008

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### Lapidary Tip

To identify and grind out pits in the cabochon, spray the rough cab with red enamel from an aerosol can, then grind the paint off with a light touch. Pits and lines will stand out as bright red spots, making it unnecessary to wipe the cab to see if the pits are out. This is especially helpful when grinding freeform cabs from Fire Agate.

Source: *The Petrified Digest* 1/99

### Tip for Silversmiths

Draw your jeweler's saw through an old candle. It coats the blade and you can saw faster. Wipe a piece of chalk over your jeweler's files. It keeps metal slivers from clogging the file and you can blow them out. Avoid scratches in silver you are working on by covering it with transparent contact paper. You can mark and saw with the paper on.

Source: *The Petrified Digest* 1/99

These tips via the *Sooner Rockologist*

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# Fiber Optic Gems: What Are They?

By Bill Grimes

Fiber optics were developed as a result of someone studying a piece of the mineral ulexite. Also known as TV stone, it is a hard, brittle, fibrous stone which-when writing is placed underneath-will allow the image to appear on the surface of the stone. This led to the theory that if this type of fibrous material could be manufactured, it could be used in many different ways where image transmission is needed. Fiber optic cables were at first very slender and flexible, used in surgeries and in household decorations. The manufacturing technology improved and soon manufacturers were spinning out miles of cable for a new application, data transmission lines. These lines can be up to two inches across. The cable consists of thousands of pairs of optic fibers. Each pair carries data for phone, computer, fax, etc. Since the sides of the cable are reflective, there is no need for insulation or shielding around each fiber, as in old phone lines. This translates to more pairs in a smaller space.

For us in the hobby, this created one of the newest gem treasures. In order to make a fiber optic cabochon, the cable scraps are first cut into small lengths. The cable is then either cut into spheres, or it is sectioned parallel to the length of the fiber, once the slices are made, it is cut much like any other gem. Care must be taken, however, to protect the ends of the cable from splintering, or catching dirt, abrasives, etc.



There is an interesting thing about fiber optic gems. If you look at them from a 90 degree angle from the eye of the gem, the gem will be transparent to light, maintaining its properties for light transmission.

Originally from *Rockhouse Roundup*, 3/99; via the *BEMS Tumbler*, 7/04; via *Golden Spike News* 8/04; via *Stratagem* 1/05 Via *The Rock Rattler* Jan 2005



One nice thing about fiber optic for us is that it is not very costly. These 40mm spheres were listed on ebay for a buy it now of \$24.99. That is for 12. Wow.

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## FRACTURE or CLEAVAGE

### Do you know the difference?

If a crystal is broken and the break is irregular, it is said to have a **fracture**.

If the break occurs along a plane and parallel to a crystal face, it has **cleavage**.

Cleavage is caused by the internal structure and varying strength of bonds between planes of different atoms, e.g. the cleavage of mica into thin sheets is called perfect cleavage.

THE CALGARY LAPIDARY JOURNAL APRIL 2011

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## WHAT'S A HEXAHEDRON ?

Don't let these long crystal names give you trouble because it's very simple if you follow and memorize a few simple rules. First, you break the word down into syllables and leave off the last syllable - **hedron**. Just remember that this means **face**. Now memorize the following five prefixes:

Tri - Three, Tetra - Four, Hexa - Six, Octa - Eight, Dodeca - Twelve.

As an example, take the word Tetra-hedron: you have hedron which means face, and Tetra which means four, Thus a Tetrahedron is a four-faced figure or crystal. Try the word Hexahedron - Hexa for six and Hedron for face: so you have a six-faced figure or crystal.

Now lets try a word with two prefixes before hedron. As an example, Tetrahexahedron. In this case, where there are two prefixes, you simply multiply the first by the second. You have tetra for four, times hexa for six and a hedron - so you have a twenty-four FACED CRYSTAL FIGURE.

Now try Hexaoctahedron: Hexa (6) x Octa (8) or a 48 faced figure. Simple isn't it ?

THE CALGARY LAPIDARY JOURNAL APRIL 2011

# Moganite

## A Common Mineral with a Disapproved Name

Mineralogists assume that they have described the common minerals of the crust. They also pride themselves on having a rigorous international system for approving new mineral names to assure the names that get into print really do represent uniquely different minerals. Recently both of these assumptions have been thrown into disarray by a mineral called "moganite". Moganite turns out to be so common that virtually every Rockhound has unknowingly found, polished or bought it.



Moganite refers to a mineral closely related to quartz. Quartz is  $\text{SiO}_2$  and crystallizes in the hexagonal system. Moganite is also  $\text{SiO}_2$ , but it crystallizes in the monoclinic system. It was discovered in 1976 by a group of geologists lead by the Russian O. W. Florke. This team was studying the rocks near Mogan in the Canary Islands and found it as fine-grained gray fibers intergrown with chalcedony and opal in cracks in the lava flows. Moganite's physical and X-ray properties are almost identical to quartz.

Florke submitted a description of moganite to the International Commission on Mineral Names and Naming (I.C.M.N.N.), who disapproved of it. They felt that not enough evidence was presented to show that moganite was not simply a mixture of quartz and other minerals. Florke and his coworkers then committed a cardinal mineralogical sin by publishing their data, using the name moganite as though it had been approved. They published more articles on moganite over the next decade, proving it was a unique mineral. They never resubmitted their data to the I.C.M.N.N. The I.C.M.N.N. just as stubbornly

refused to approve the name without such an official resubmission. So goes mineralogy politics!

A recent article by Peter Heaney and Jeffrey Post in Science added insult to injury. They reported finding moganite in nearly all of the 150 samples of fine-grained quartz they tested from the Smithsonian's mineral collection. Some samples contained over 75% moganite mixed with quartz. Chert had the most moganite. Flint contained 13-17% moganite. Agate had 5 to 20% moganite. Silicified corals from the Tampa Bay area in Florida averaged 20% moganite. The only samples found with no moganite were jaspers from iron formations, the weathered outer rinds of agates and Arkansas novaculite. Clearly moganite is a very overlooked mineral.

Why moganite forms is not known. In the Canary Islands it forms under extremely dry surface conditions. Since moganite is more soluble than quartz, it leaches out of chalcedony, which could account for chalcedony's lower density and higher permeability.

Here is a mineral we all have that doesn't have an approved name. What is a conscientious mineral collector to do? We'd best just label things "quartz", "agate" or "chalcedony" until the semantic dust settles.

Dr. Bill Cordua, University of Wisconsin – River Falls.

### References:

Florke, O.W., U. Florke and U. Giese, 1984, "Moganite: a new microcrystalline silica mineral", *Nues Jahrb. Mineralogy*, vol. 149, p. 325-336.

Heaney, Peter and Post, J.E., 1992, "The widespread distribution of novel silica polymorph in microcrystalline quartz", *Science*, vol. 255, p. 441-444.

Via: *Rock Chatter* Publication Sept 2011



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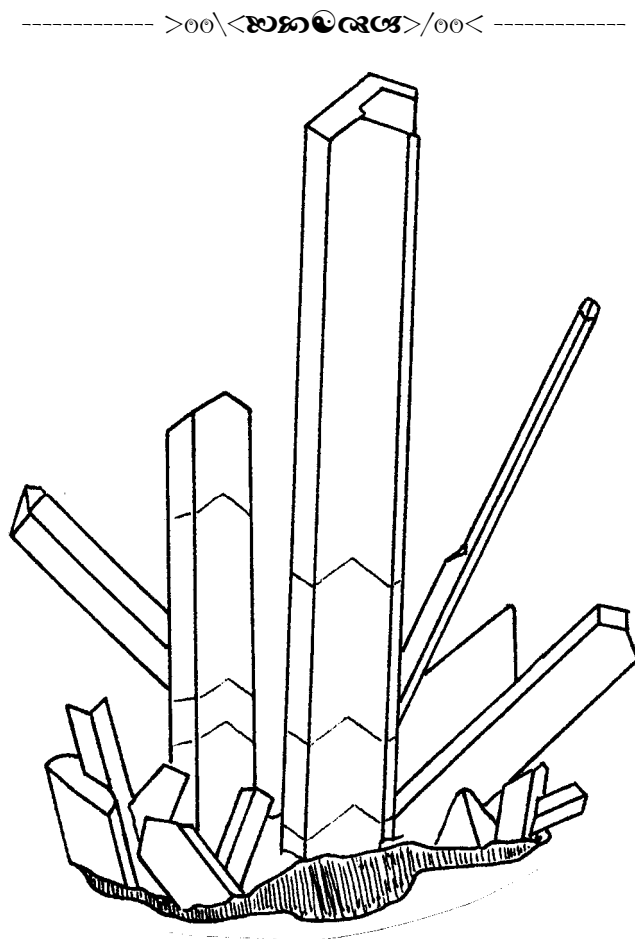
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